

Solar Eclipses: Front Line Experiences, and Some Science

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NASA/MSFC

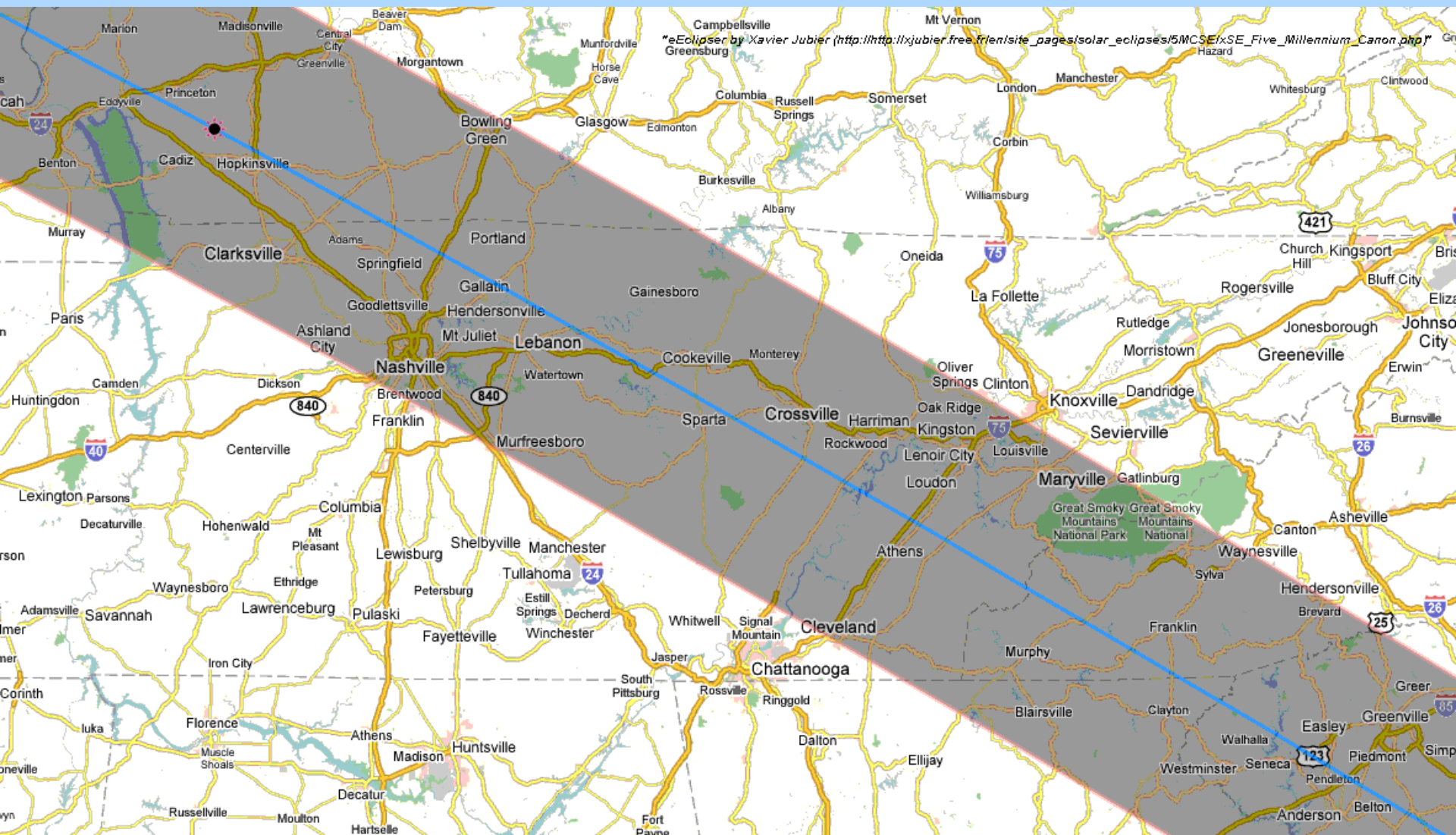
Today's Discussion:

- A brief review of the 21 August 2017 eclipse.
- A couple of experiences from the “front line” of past eclipses.
- The motivation for doing solar eclipse studies (just an outline...).

August 21, 2017 Total Solar Eclipse Path



August 21, 2017 Total Solar Eclipse Path



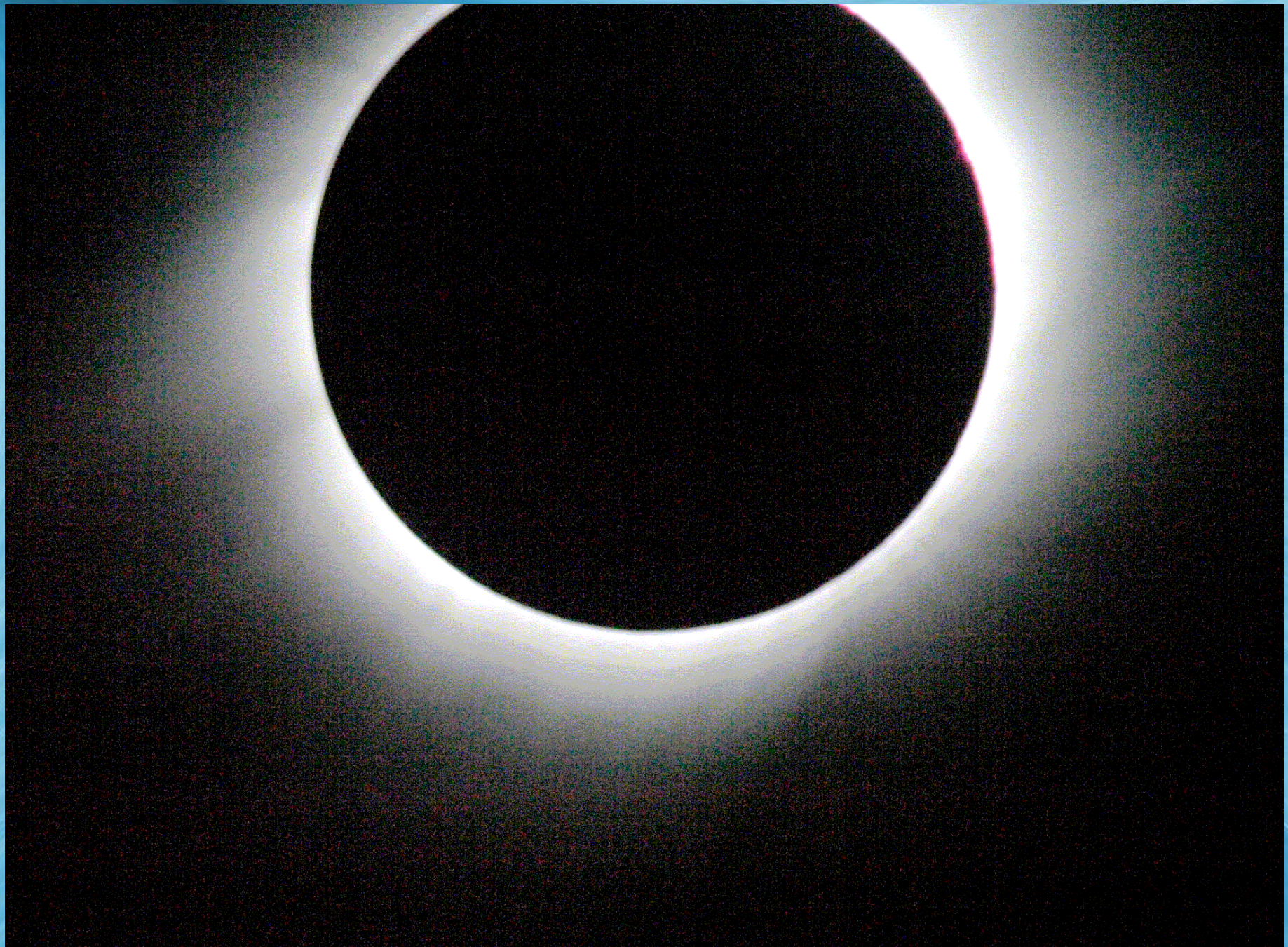
Do not Risk Your Precious Eyes!!

- Do ***not*** look at the partial phases of the eclipse directly without certified eye protection!!
- There's no point to staring at partial phases....
- Look without protection only if you're in the totality path, and then only during totality! (About 2 minutes.)

How *Not* to do Eclipse Science! Ghana 2006 Version

(Useful information for picture takers too.)





Lessons Learned (Science at Eclipses)

- 🕒 No new equipment.
- 🕒 Settle on a location, and stick to it if at all possible.
- 🕒 Practice, practice, practice!

Gansu Province, China, 2008











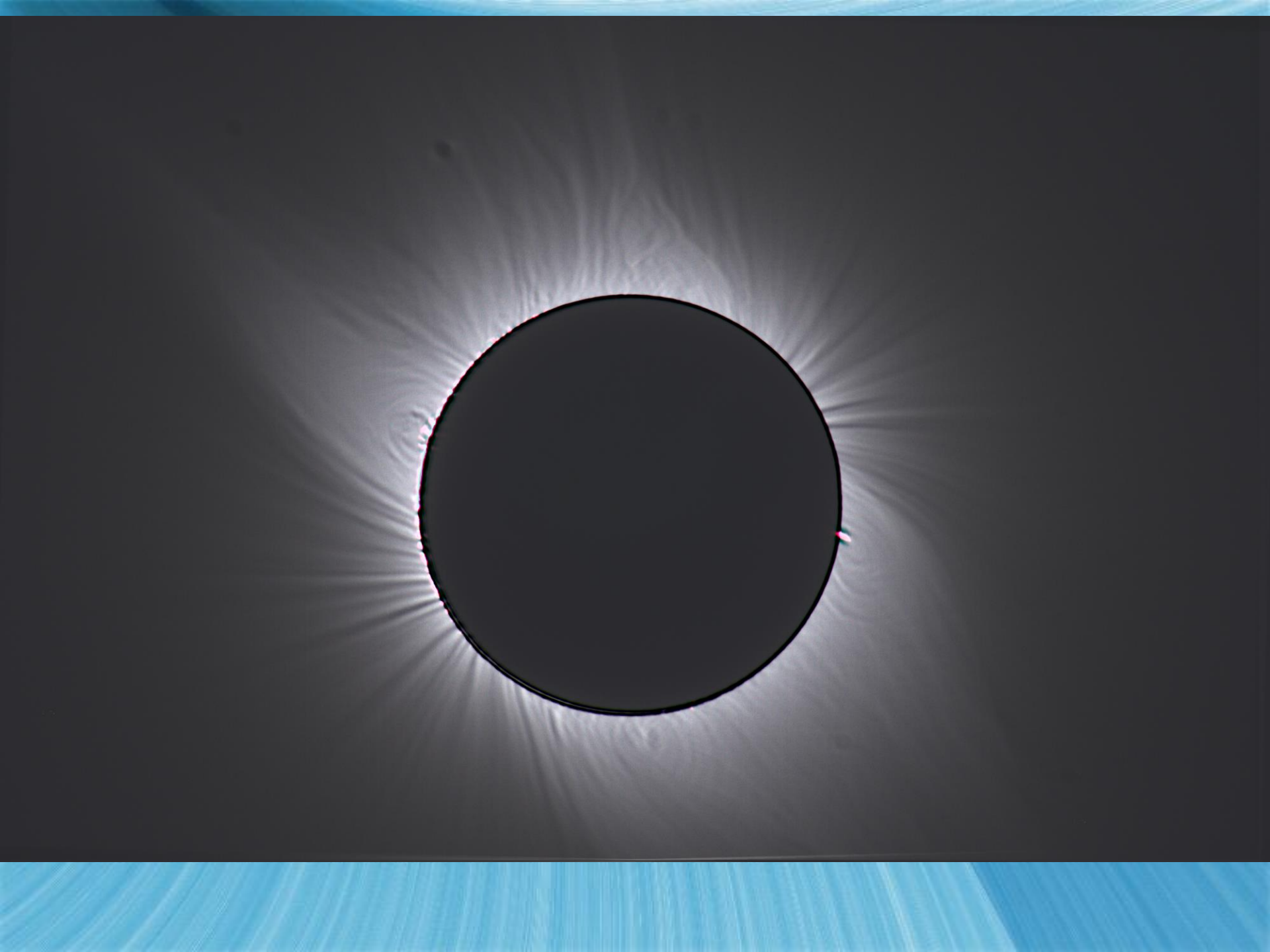




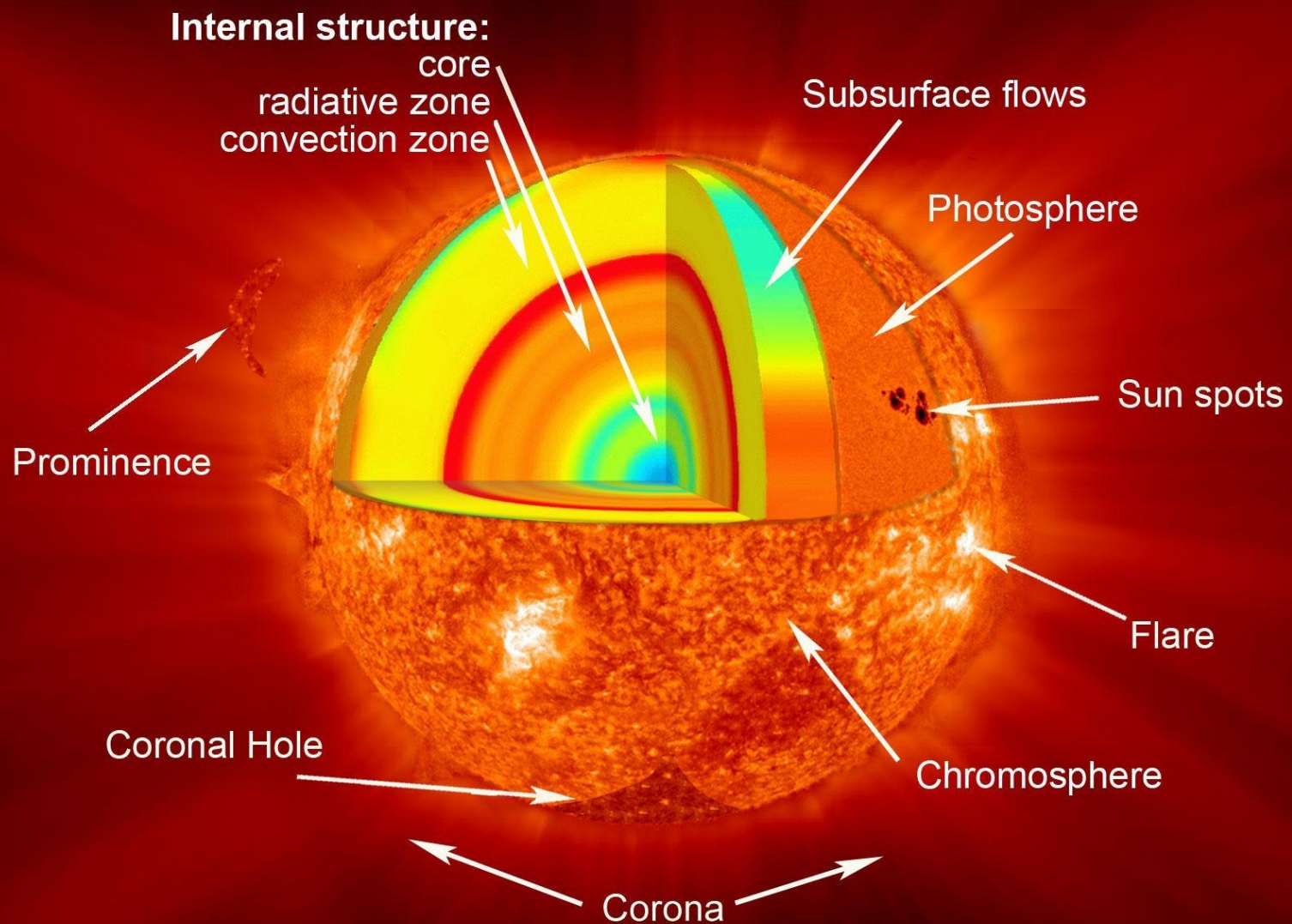


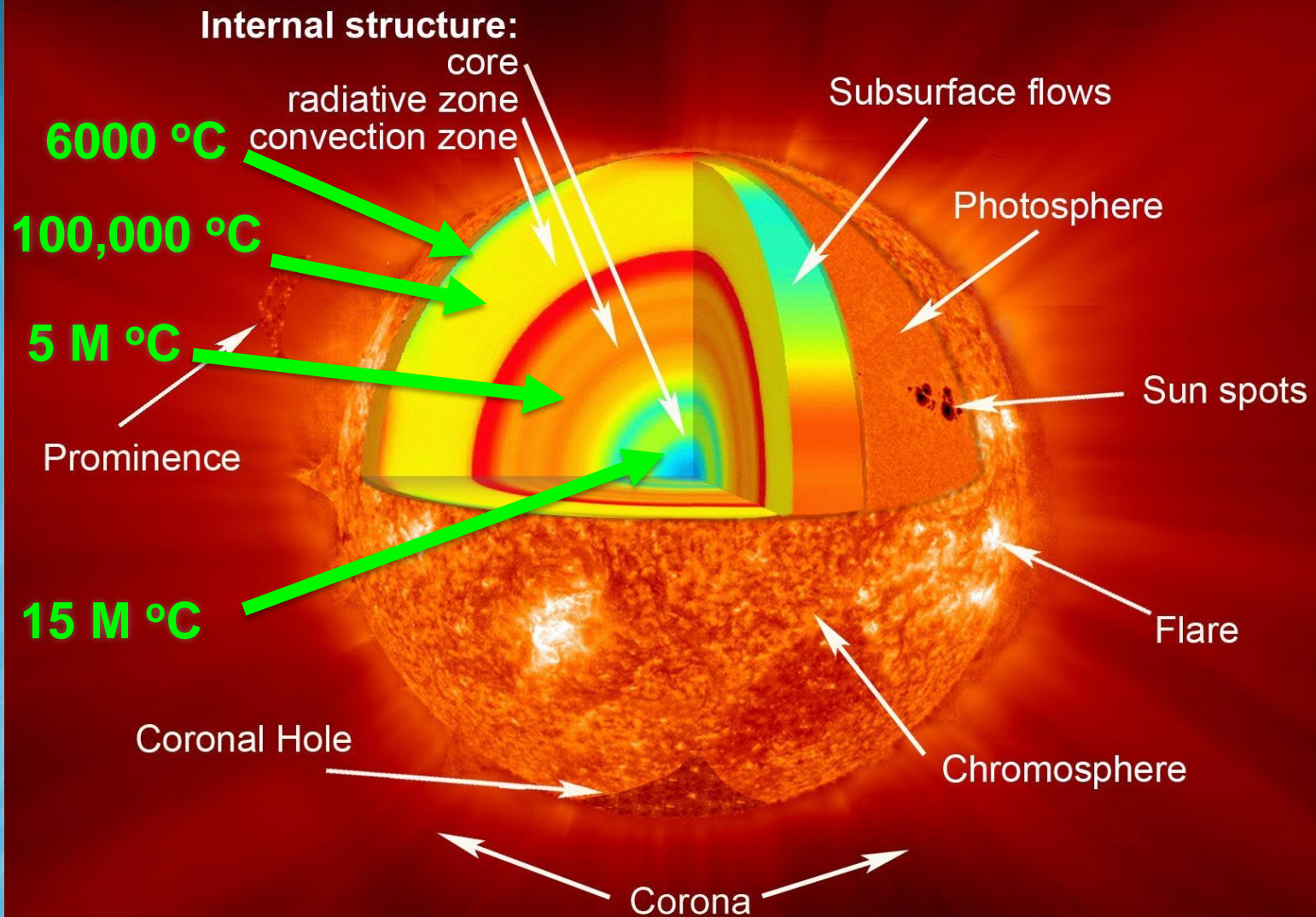






Science of the Sun. A key Problem: The Temperature of the Corona

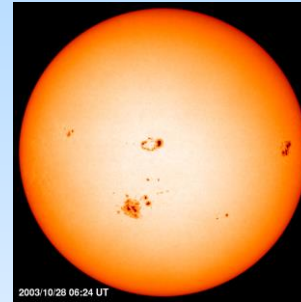




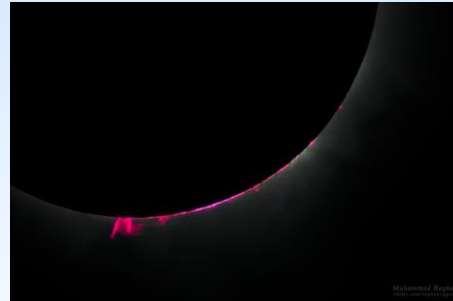
The Solar Atmosphere

The Outer layers (Atmospheres) of the Sun:

- Photosphere



- Chromosphere



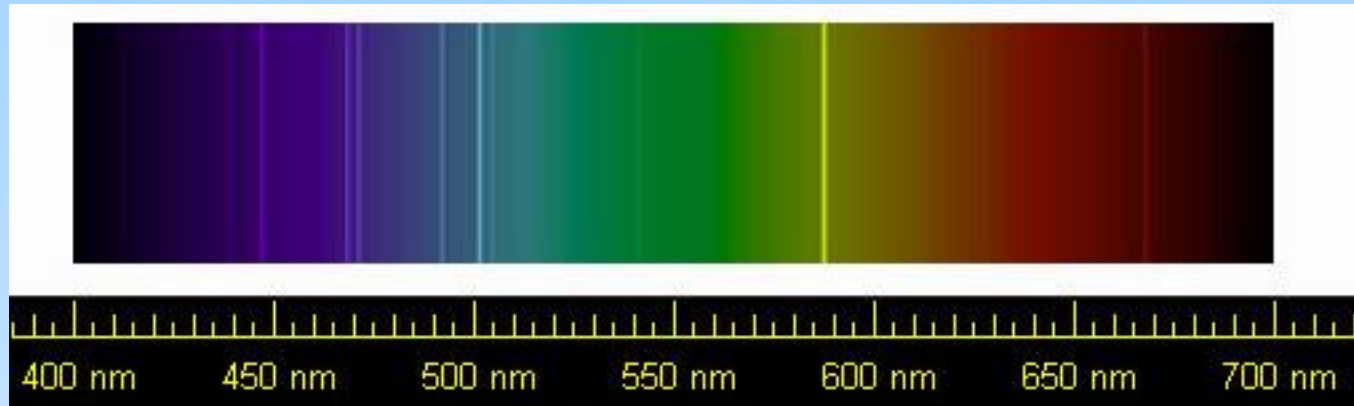
- Corona



But....

- A “problem” resulted from 1869 eclipse observations of the corona.
- Used a recently developed instrument - the “spectroscope,” to study the corona.

Spectra



- Breaks light into “rainbow” colors, with lines.
- Different elements produce different lines.
- So these lines are like “fingerprints” for elements.



- Hence, spectra can be used to determine the composition of far-away objects, like the corona.

But...

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- Used a recently developed instrument - the “spectroscope,” to study the corona.
- Found lines that could not be identified!

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- A “problem” resulted from 1869 eclipse observations of the corona.
- Used a recently developed instrument - the “spectroscope,” to study the corona.
- Found lines that could not be identified!
- Many explanations considered, including a “new” element: *coronium*.

But this didn't work....

The Periodic Table

<div><div><div>1</div><div>H</div><div>hydrogen</div><div>1.007 94(7)</div></div><div><div>Atomic Number</div><div>Symbol</div><div>Name</div><div>Standard Atomic Weight</div></div></div>																																			
<div><div><div>Metals</div><div>Transition Metals</div><div>Metalloids</div></div><div><div>Non-metals</div><div>Lanthanoids</div><div>Halogens</div></div><div><div>Alkali Metals</div><div>Actinoids</div><div>Noble Gases</div></div><div><div>Alkali Earth Metals</div></div></div>																																			
<div><div><div>3</div><div>Li</div><div>lithium</div><div>6.941(3)</div></div><div><div>4</div><div>Be</div><div>beryllium</div><div>9.012 183(2)</div></div><div><div>5</div><div>B</div><div>boron</div><div>10.811(7)</div></div><div><div>6</div><div>C</div><div>carbon</div><div>12.010 7(8)</div></div><div><div>7</div><div>N</div><div>nitrogen</div><div>14.006 4(1)</div></div><div><div>8</div><div>O</div><div>oxygen</div><div>15.999 4(3)</div></div><div><div>9</div><div>F</div><div>fluorine</div><div>18.998 403(1)</div></div><div><div>10</div><div>Ne</div><div>neon</div><div>20.179 7(4)</div></div></div>																																			
<div><div><div>11</div><div>Na</div><div>sodium</div><div>22.989 769(2)</div></div><div><div>12</div><div>Mg</div><div>magnesium</div><div>24.304 0(4)</div></div><div><div>13</div><div>Al</div><div>aluminum</div><div>26.981 538(6)</div></div><div><div>14</div><div>Si</div><div>silicon</div><div>28.085 5(3)</div></div><div><div>15</div><div>P</div><div>phosphorus</div><div>30.973 762(2)</div></div><div><div>16</div><div>S</div><div>sulfur</div><div>32.06(5)</div></div><div><div>17</div><div>Cl</div><div>chlorine</div><div>35.45(3)</div></div><div><div>18</div><div>Ar</div><div>argon</div><div>39.948(1)</div></div></div>																																			
<div><div><div>19</div><div>K</div><div>potassium</div><div>39.098 3(1)</div></div><div><div>20</div><div>Ca</div><div>calcium</div><div>40.078 4(2)</div></div><div><div>21</div><div>Sc</div><div>scandium</div><div>44.955 912(6)</div></div><div><div>22</div><div>Ti</div><div>titanium</div><div>47.88(7)</div></div><div><div>23</div><div>V</div><div>vanadium</div><div>50.941 5(1)</div></div><div><div>24</div><div>Cr</div><div>chromium</div><div>51.996 1(5)</div></div><div><div>25</div><div>Mn</div><div>manganese</div><div>54.938 045(3)</div></div><div><div>26</div><div>Fe</div><div>iron</div><div>55.845(2)</div></div><div><div>27</div><div>Co</div><div>cobalt</div><div>58.933 194(5)</div></div><div><div>28</div><div>Ni</div><div>nickel</div><div>58.693 4(2)</div></div><div><div>29</div><div>Cu</div><div>copper</div><div>63.546(3)</div></div><div><div>30</div><div>Zn</div><div>zinc</div><div>65.38(4)</div></div><div><div>31</div><div>Ga</div><div>gallium</div><div>69.723 1(2)</div></div><div><div>32</div><div>Ge</div><div>germanium</div><div>72.64(1)</div></div><div><div>33</div><div>As</div><div>arsenic</div><div>74.921 60(2)</div></div><div><div>34</div><div>Se</div><div>selenium</div><div>78.96(4)</div></div><div><div>35</div><div>Br</div><div>bromine</div><div>79.904(1)</div></div><div><div>36</div><div>Kr</div><div>krypton</div><div>83.80(1)</div></div></div>																																			
<div><div><div>37</div><div>Rb</div><div>rubidium</div><div>85.467 8(2)</div></div><div><div>38</div><div>Sr</div><div>strontium</div><div>87.62(1)</div></div><div><div>39</div><div>Y</div><div>yttrium</div><div>88.905 85(2)</div></div><div><div>40</div><div>Zr</div><div>zirconium</div><div>91.224(2)</div></div><div><div>41</div><div>Nb</div><div>niobium</div><div>92.906 38(2)</div></div><div><div>42</div><div>Mo</div><div>molybdenum</div><div>95.94(2)</div></div><div><div>43</div><div>Tc</div><div>technetium</div><div>[98]</div></div><div><div>44</div><div>Ru</div><div>ruthenium</div><div>101.07(2)</div></div><div><div>45</div><div>Rh</div><div>rhodium</div><div>102.905 50(2)</div></div><div><div>46</div><div>Pd</div><div>palladium</div><div>106.42(2)</div></div><div><div>47</div><div>Ag</div><div>silver</div><div>107.868 2(2)</div></div><div><div>48</div><div>Cd</div><div>cadmium</div><div>112.411 5(2)</div></div><div><div>49</div><div>In</div><div>indium</div><div>114.818 1(2)</div></div><div><div>50</div><div>Sn</div><div>tin</div><div>118.710(7)</div></div><div><div>51</div><div>Sb</div><div>antimony</div><div>121.757(1)</div></div><div><div>52</div><div>Te</div><div>tellurium</div><div>127.60(3)</div></div><div><div>53</div><div>I</div><div>iodine</div><div>126.905 4(5)</div></div><div><div>54</div><div>Xe</div><div>xenon</div><div>131.29(4)</div></div></div>																																			
<div><div><div>55</div><div>Cs</div><div>cesium</div><div>132.905 451(3)</div></div><div><div>56</div><div>Ba</div><div>barium</div><div>137.327(7)</div></div><div><div>57-71</div><div>La-Lu</div><div>lanthanoids</div></div><div><div>72</div><div>Hf</div><div>hafnium</div><div>178.49(2)</div></div><div><div>73</div><div>Ta</div><div>tantalum</div><div>180.947 3(2)</div></div><div><div>74</div><div>W</div><div>tungsten</div><div>183.84(1)</div></div><div><div>75</div><div>Re</div><div>rhenium</div><div>186.207(1)</div></div><div><div>76</div><div>Os</div><div>osmium</div><div>190.23(3)</div></div><div><div>77</div><div>Ir</div><div>iridium</div><div>192.222(1)</div></div><div><div>78</div><div>Pt</div><div>platinum</div><div>195.084(9)</div></div><div><div>79</div><div>Au</div><div>gold</div><div>196.966 569(4)</div></div><div><div>80</div><div>Hg</div><div>mercury</div><div>200.59(2)</div></div><div><div>81</div><div>Tl</div><div>thallium</div><div>204.383 3(2)</div></div><div><div>82</div><div>Pb</div><div>lead</div><div>207.2(1)</div></div><div><div>83</div><div>Bi</div><div>bismuth</div><div>208.980 4(1)</div></div><div><div>84</div><div>Po</div><div>polonium</div><div>[209]</div></div><div><div>85</div><div>At</div><div>astatine</div><div>[210]</div></div><div><div>86</div><div>Rn</div><div>radon</div><div>[222]</div></div></div>																																			
<div><div><div>87</div><div>Fr</div><div>francium</div><div>[223]</div></div><div><div>88</div><div>Ra</div><div>radium</div><div>[226]</div></div><div><div>89-103</div><div>Ac-Lr</div><div>actinoids</div></div><div><div>104</div><div>Rf</div><div>rutherfordium</div><div>[261]</div></div><div><div>105</div><div>Db</div><div>dubnium</div><div>[262]</div></div><div><div>106</div><div>Sg</div><div>seaborgium</div><div>[266]</div></div><div><div>107</div><div>Bh</div><div>bohrium</div><div>[264]</div></div><div><div>108</div><div>Hs</div><div>hassium</div><div>[277]</div></div><div><div>109</div><div>Mt</div><div>meitnerium</div><div>[268]</div></div><div><div>110</div><div>Ds</div><div>darmstadtium</div><div>[271]</div></div><div><div>111</div><div>Rg</div><div>roentgenium</div><div>[272]</div></div></div>																																			
<div><div><div>lanthanoids</div><div>57</div><div>La</div><div>lanthanum</div><div>138.905 47(7)</div></div><div><div>58</div><div>Ce</div><div>cerium</div><div>140.12(1)</div></div><div><div>59</div><div>Pr</div><div>praseodymium</div><div>140.907 65(2)</div></div><div><div>60</div><div>Nd</div><div>neodymium</div><div>144.24(2)</div></div><div><div>61</div><div>Pm</div><div>promethium</div><div>[145]</div></div><div><div>62</div><div>Sm</div><div>samarium</div><div>150.36(2)</div></div><div><div>63</div><div>Eu</div><div>euroium</div><div>151.964(1)</div></div><div><div>64</div><div>Gd</div><div>gadolinium</div><div>157.25(2)</div></div><div><div>65</div><div>Tb</div><div>terbium</div><div>158.925 3(2)</div></div><div><div>66</div><div>Dy</div><div>dysprosium</div><div>162.500 1(2)</div></div><div><div>67</div><div>Ho</div><div>holmium</div><div>164.930 32(2)</div></div><div><div>68</div><div>Er</div><div>erbium</div><div>167.259(3)</div></div><div><div>69</div><div>Tm</div><div>thulium</div><div>168.930 28(2)</div></div><div><div>70</div><div>Yb</div><div>ytterbium</div><div>173.054 7(1)</div></div><div><div>71</div><div>Lu</div><div>lutetium</div><div>174.967(1)</div></div></div>																																			
<div><div><div>actinoids</div><div>89</div><div>Ac</div><div>actinium</div><div>[227]</div></div><div><div>90</div><div>Th</div><div>thorium</div><div>232.037 7(4)</div></div><div><div>91</div><div>Pa</div><div>protactinium</div><div>231.036 88(1)</div></div><div><div>92</div><div>U</div><div>uranium</div><div>238.028 91(3)</div></div><div><div>93</div><div>Np</div><div>neptunium</div><div>[237]</div></div><div><div>94</div><div>Pu</div><div>plutonium</div><div>[244]</div></div><div><div>95</div><div>Am</div><div>americium</div><div>[243]</div></div><div><div>96</div><div>Cm</div><div>curium</div><div>[247]</div></div><div><div>97</div><div>Bk</div><div>berkelium</div><div>[247]</div></div><div><div>98</div><div>Cf</div><div>californium</div><div>[251]</div></div><div><div>99</div><div>Es</div><div>einsteinium</div><div>[252]</div></div><div><div>100</div><div>Fm</div><div>fermium</div><div>[257]</div></div><div><div>101</div><div>Md</div><div>mendelevium</div><div>[258]</div></div><div><div>102</div><div>No</div><div>nobelium</div><div>[259]</div></div><div><div>103</div><div>Lr</div><div>lawrencium</div><div>[262]</div></div></div>																																			

NASA/CXC/SAO

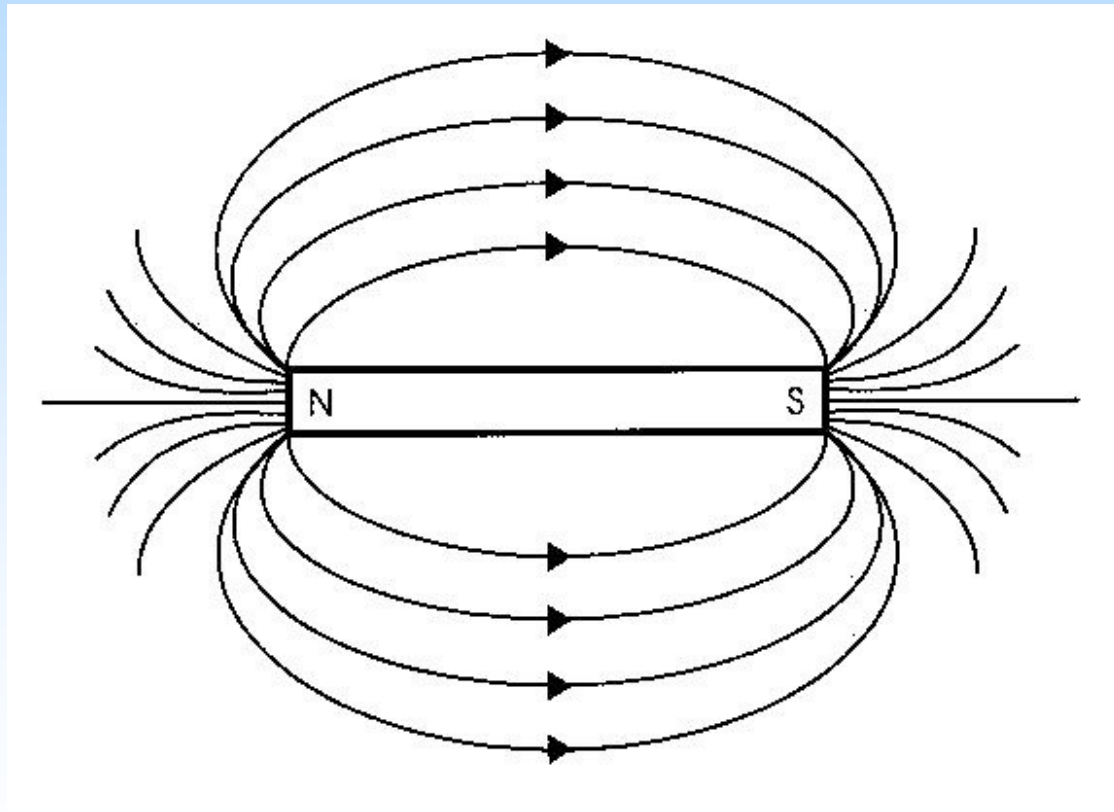
The Corona: Continued...

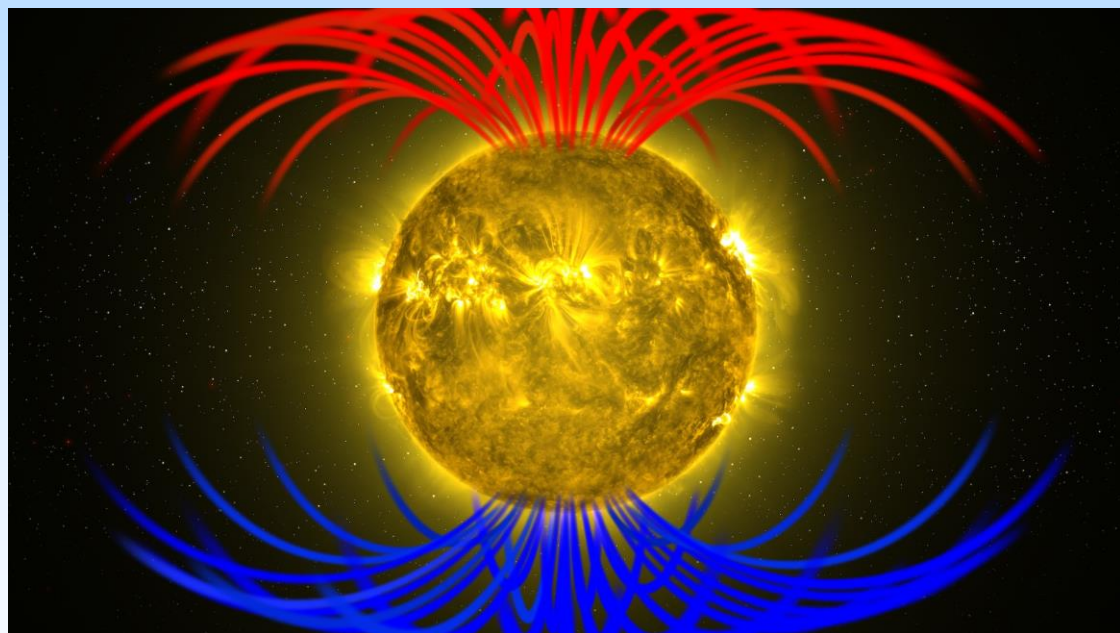
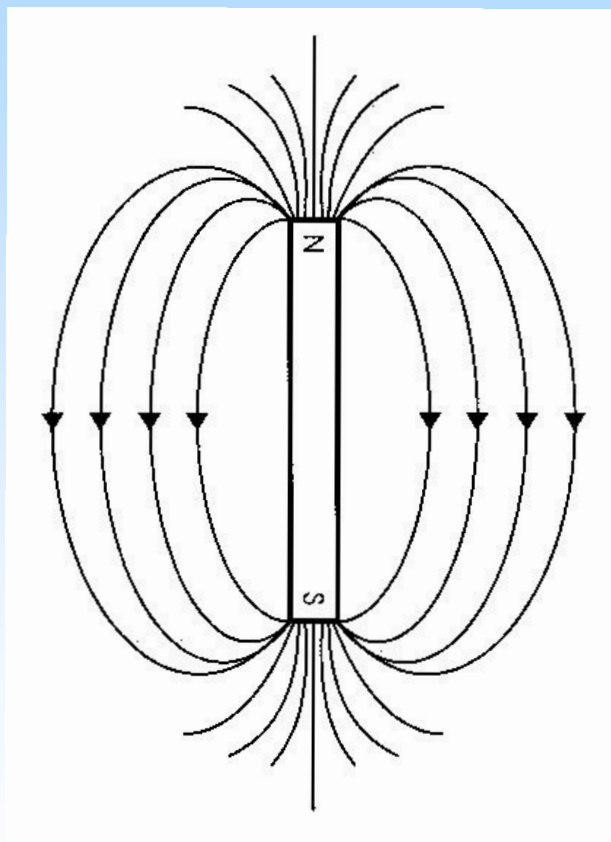
- The mystery spectral lines found to be due to very hot (“highly-ionized”) familiar elements ~1940.

So this was a slooooow process: 1869 eclipse observations, and 1939~1943 explanation!!

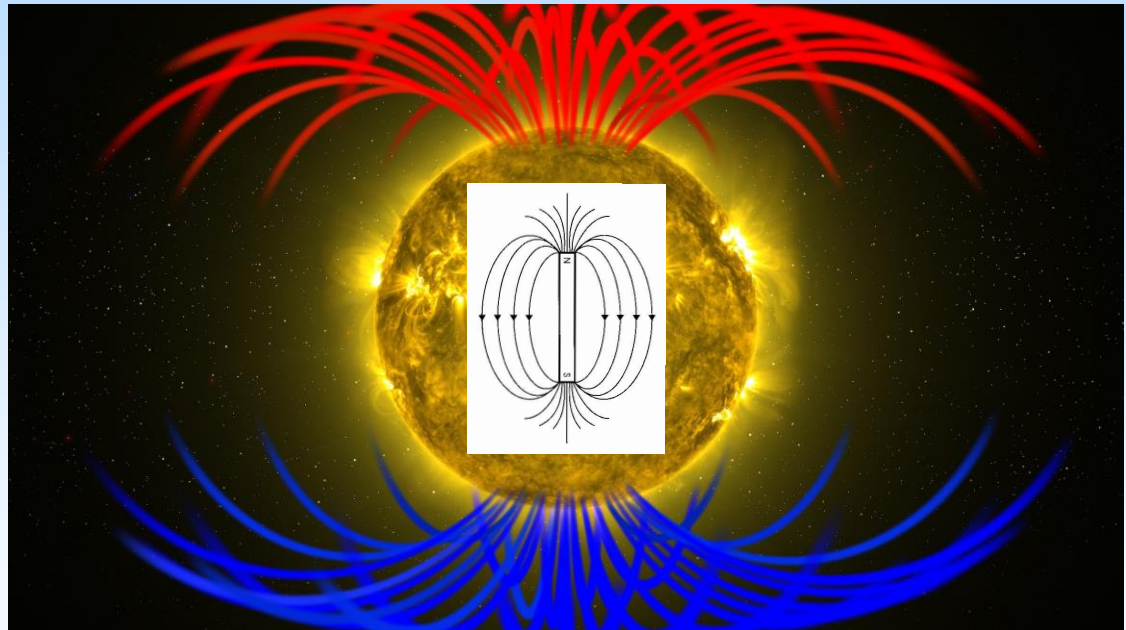
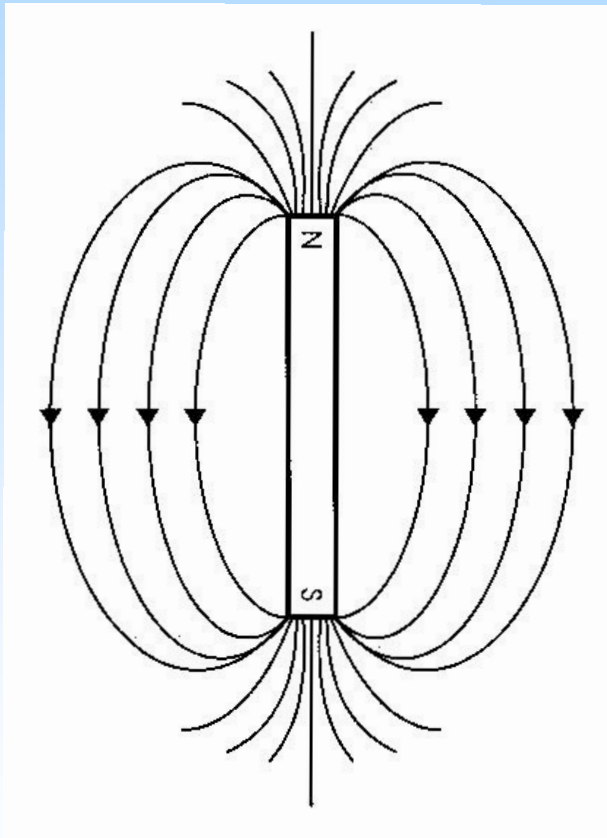
- Structure of the corona: late 1960s and 1970s observations from balloons, Skylab, etc.
- This structure due to the magnetic field.

Magnetism is the key to why the corona is hot. (It is also responsible for many of the changing features of the Sun.)

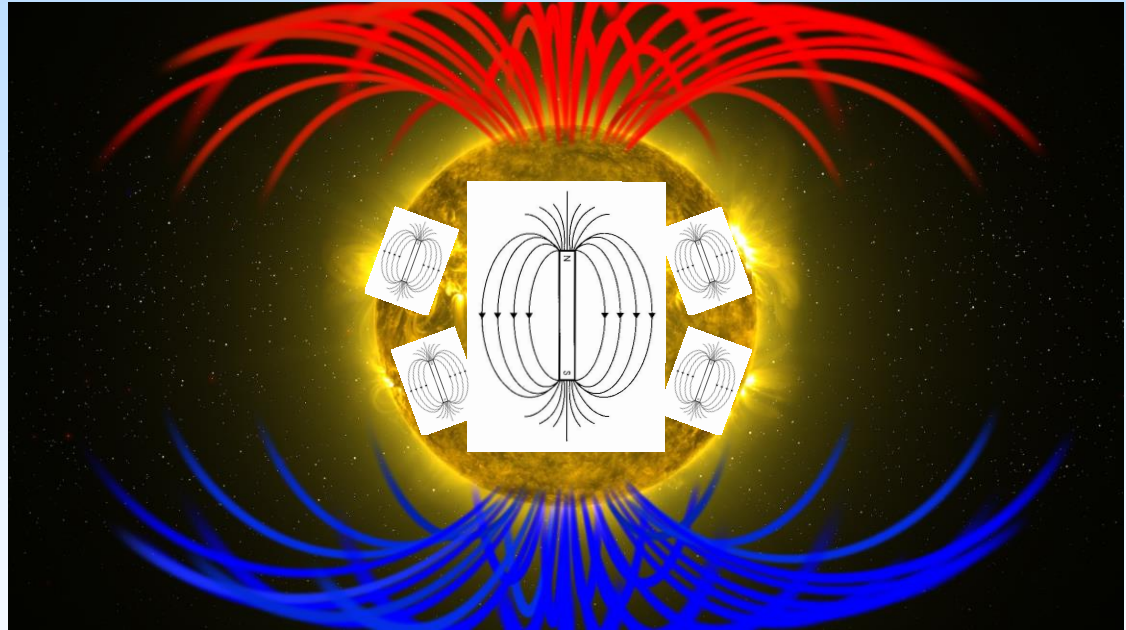
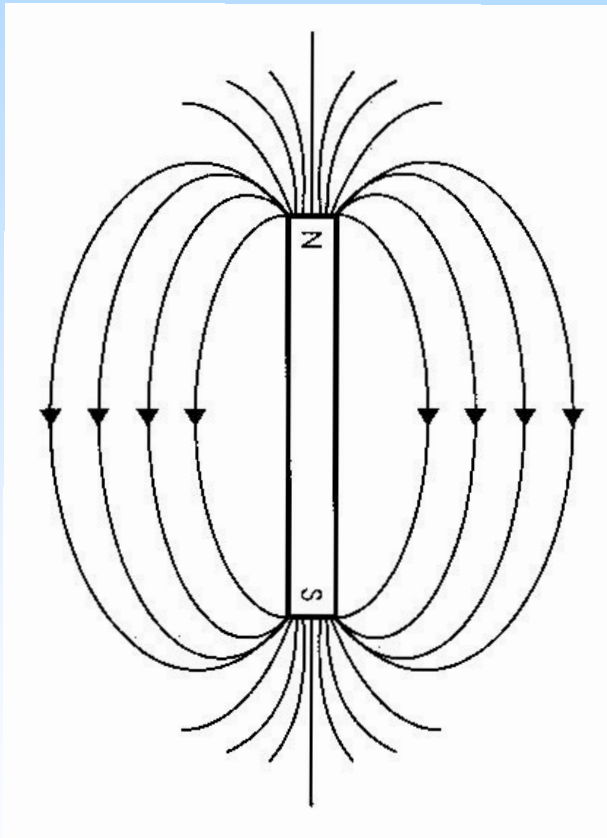


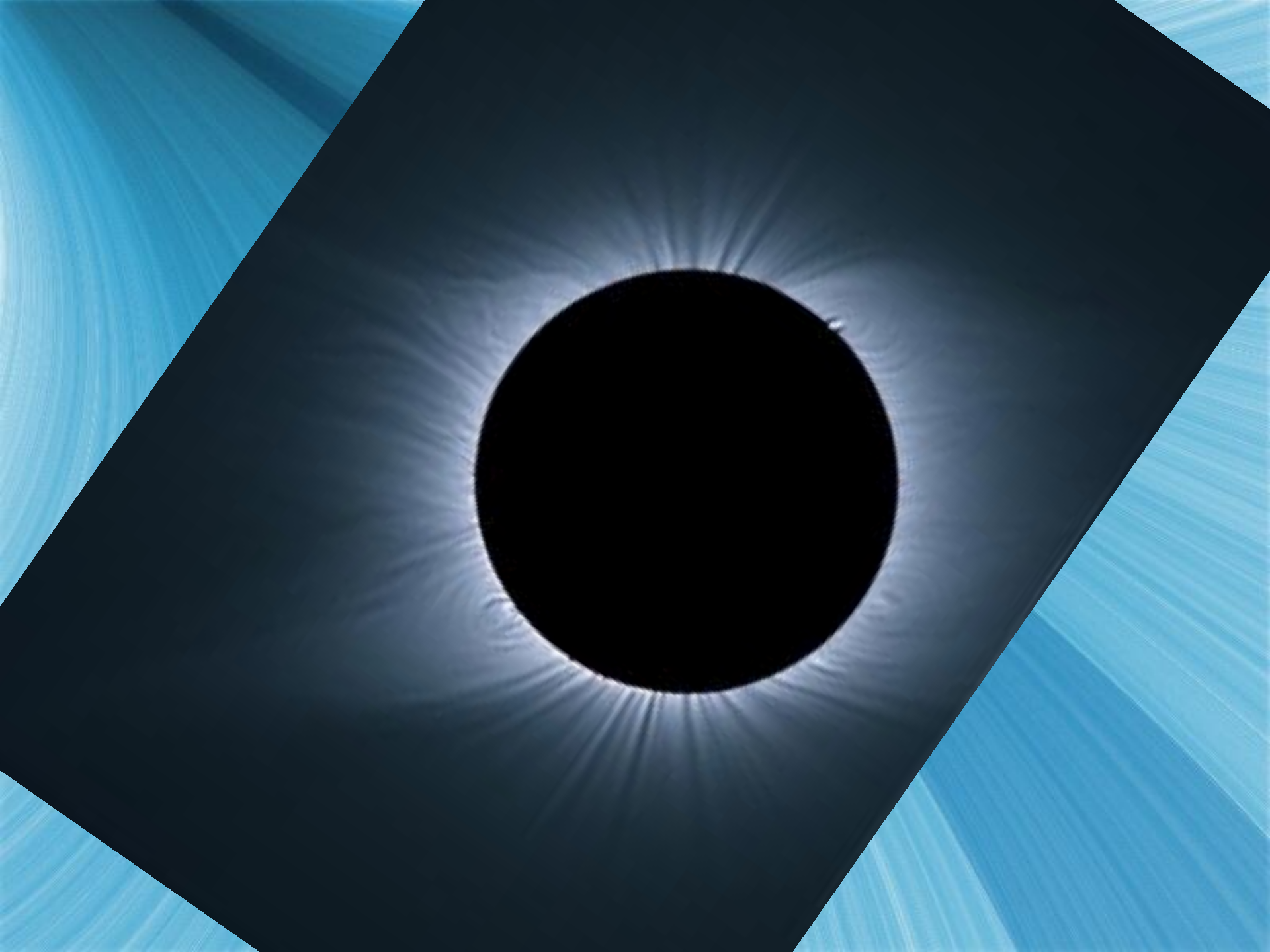


Magnetism is the key to why the corona is hot.



Magnetism is the key to why the corona is hot.





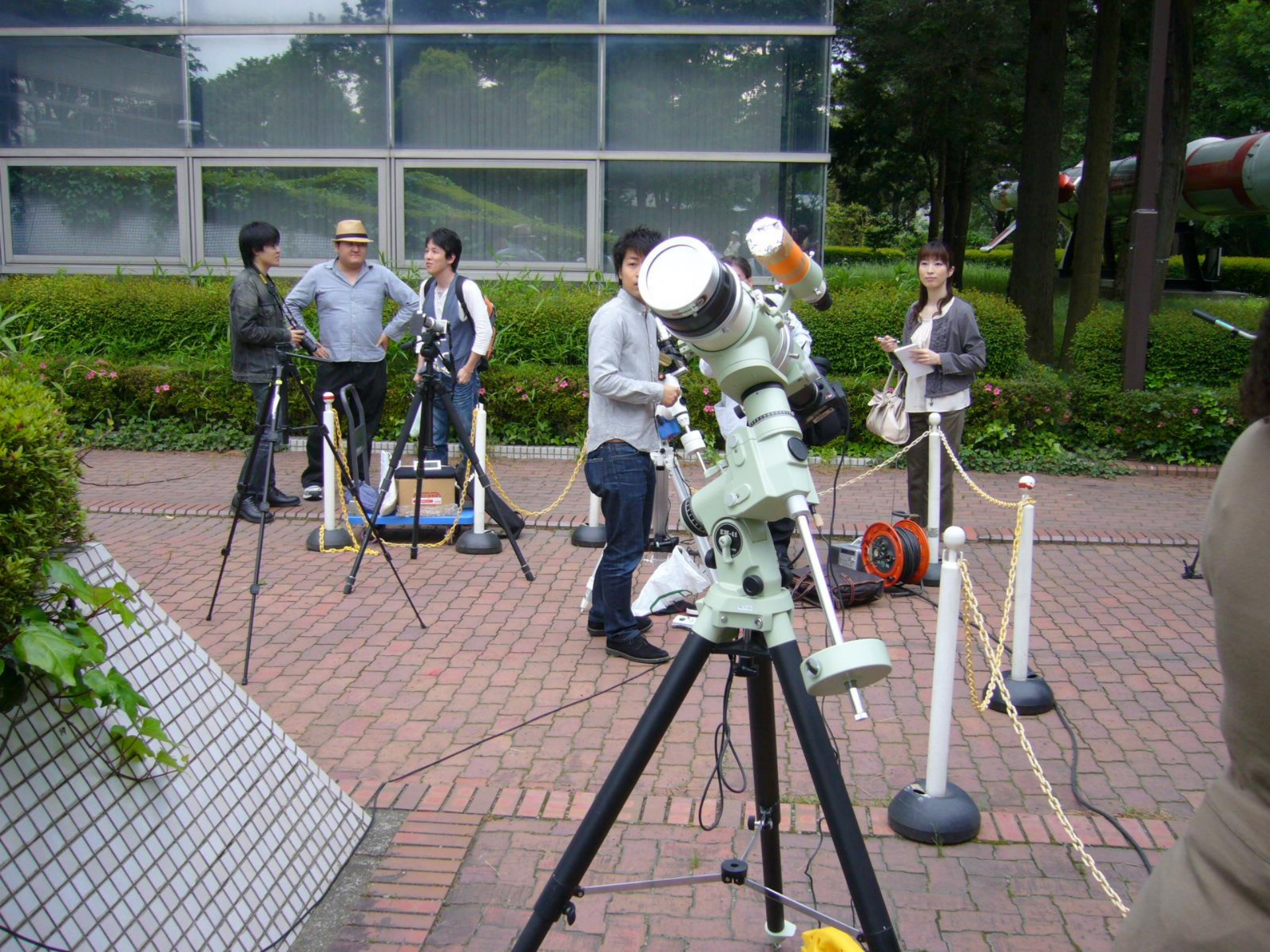
A Key Objective of Solar Physics:

Unravel details of how the magnetic field heats the corona!

- This is one goal (direct or indirect) of many eclipse studies.
- Also, many other stars have hot coronae, and thus understanding the heating of the Sun's corona tells us about other stars too.











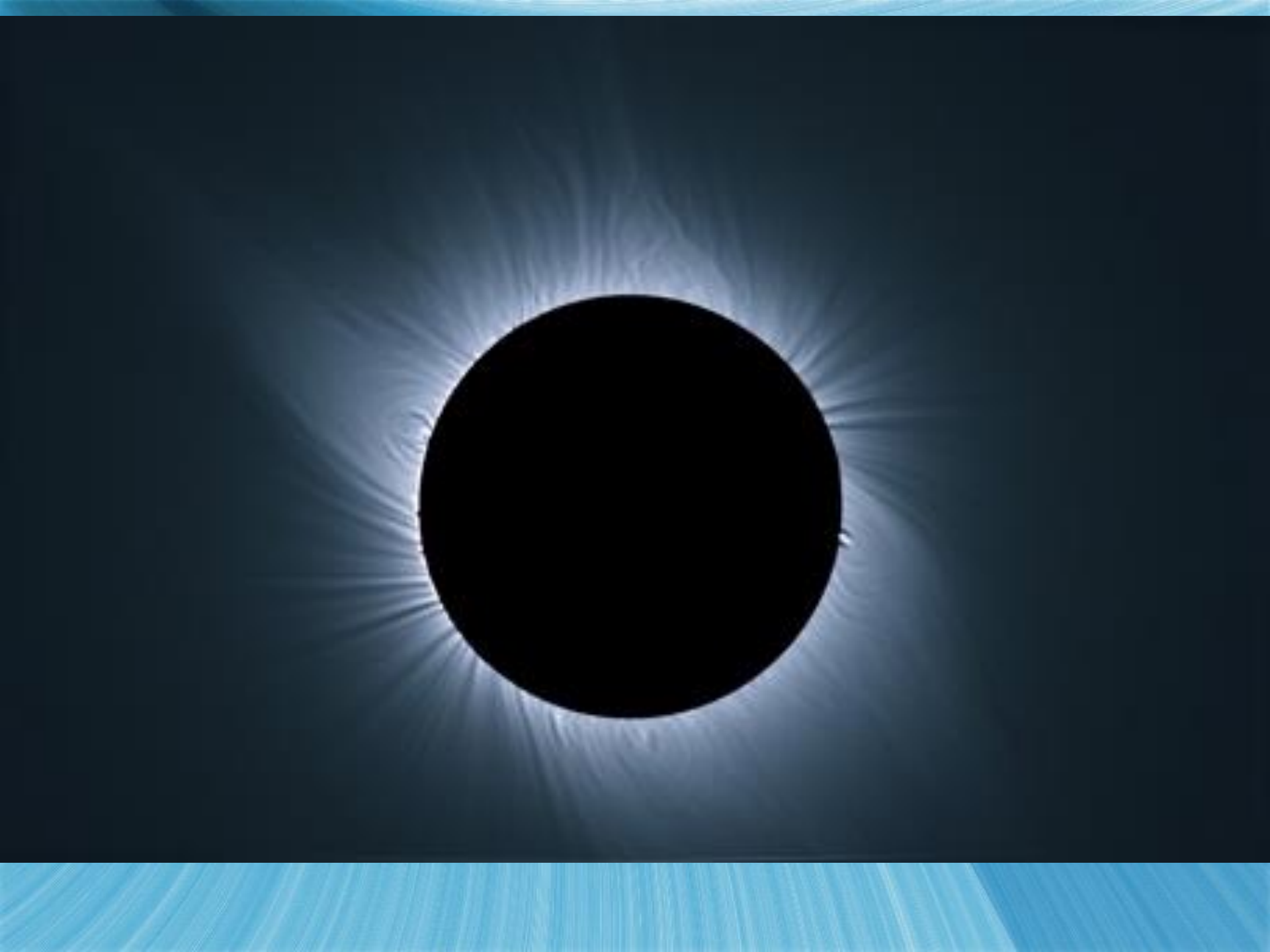
20 May 2012 22:35:02 UT



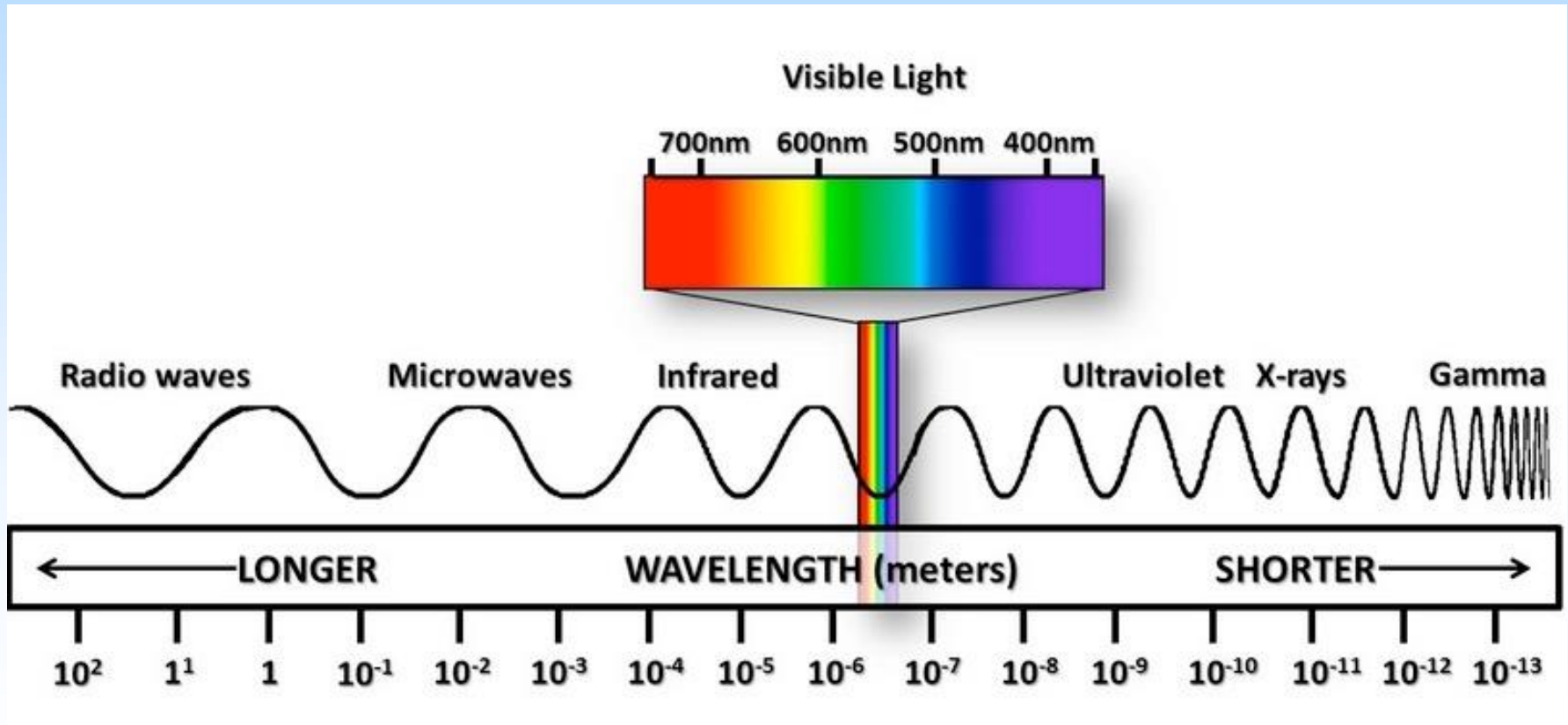
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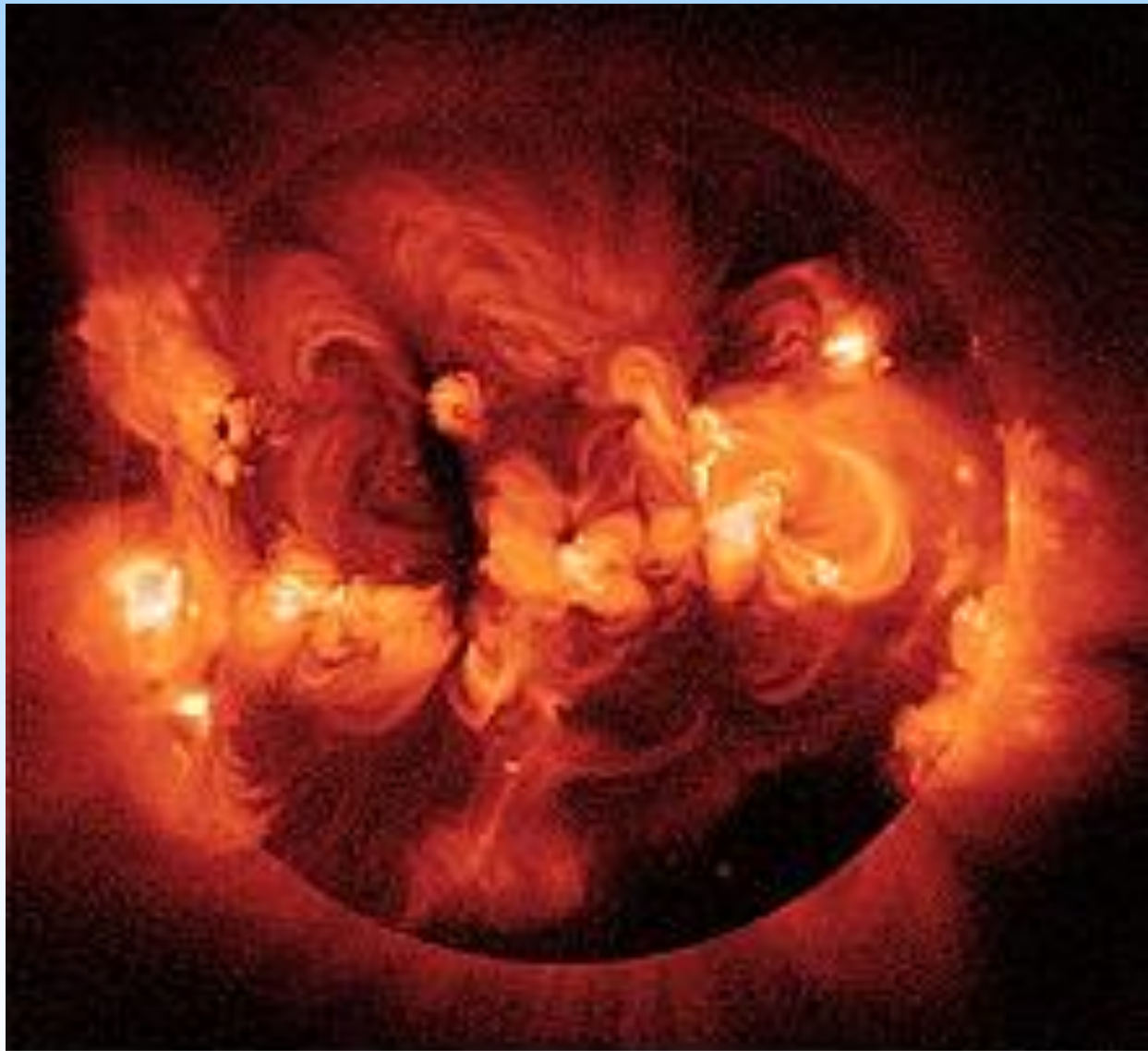






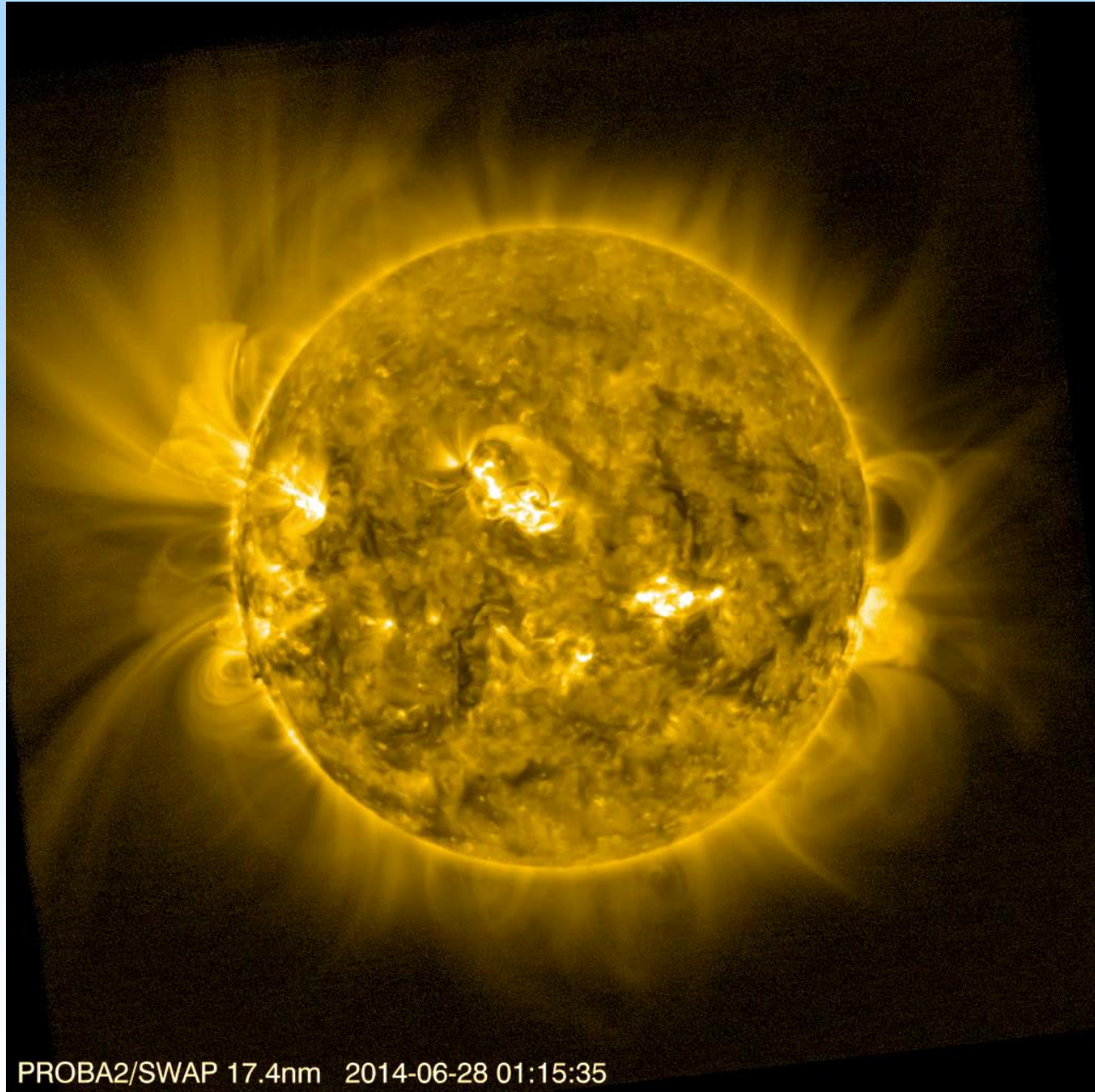
We have to go to *space* to see the Sun's outer atmosphere with regularity.





NASA

The Corona from Yohkoh/SXT



PROBA2/SWAP 17.4nm 2014-06-28 01:15:35

The Corona

- Expected to be cool, but found strange spectral lines, first during 1869 eclipse.
- Many explanations considered, including a “new” element: *coronium*.
- **But this didn't work....**